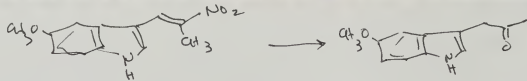


(4)

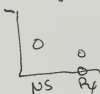


$\left. \begin{array}{l} 8 \text{ ml H}_2\text{O} \\ 2 \text{ ml H}_2\text{O} \end{array} \right\} \Delta \text{ on S.B. add}$

1.7 g electrolytic iron dust, when hot, add
0.5 g NS a glob at a time, over 2 minutes

10 min all NS gone - new spots at origin
& in between

TLC
CH₂Cl₂
hexane



30 min off (had to wait for TLC)
and CH₂Cl₂

Into 100 ml H₂O - 2 phases and secondary in between -
filter through paper - separate, 2x50 ml CH₂Cl₂ more - wash
̄ sat K₂CO₃ - flash → 0.28 g film.

KR at 0.1 mm.

130°

160

180° (vac to 0.08)

215-230° over → 0.24 g oil - off white

IR C=O at 1710 cm⁻¹ mass spec OK.

stand a couple of months → deep brown color.

Repeat:

$\left. \begin{array}{l} 20 \text{ ml H}_2\text{O} \\ 5 \text{ ml H}_2\text{O} \end{array} \right\} \Delta \text{ SB } \sim 60^\circ \text{ add}$

4.55 g Fe electrolytic - up to ~ 70° - bubbling. add

1.24 g nitrostyrene - over 2 minutes - color disappears
130 W

5 minutes OFF. - into 250 H₂O, 50 ml CH₂Cl₂

Filter through paper - wash ̄ CH₂Cl₂. separate - xht ag.

̄ 2x50 CH₂Cl₂ - combine, flash → 1.17 g amber oil - slight
acetic acid smell

Stand 3 days in cold lab → xhals! mp - not yet by 95° - polymer?

8:4

TLC (CH₂Cl₂ 2:

OK. hot
however.